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## ABSTRACT

There is a continuing need to increase the velocity and associated terminal performance of kinetic energy ammunition as tougher armor targets are encountered. Application of a deterrent, or burning rate reducer, into the surface of standard propellants, together with appropriate changes in geometry and loading density, has been suggested as a means of increasing velocity by as much as ten percent. The attainment of such a desirable performance gain is critically dependent on the deterrent's satisfying certain requirements: (1) chemical compatibility with the base propellant, (2) sufficient penetration of the base propellant, (3) diffusion stability of the deterred region over an extended time, (4) ignitability of the deterred layer, (5) sufficient decrease in burning rate. Unfortunately, there exists only a minuscule data base on the way deterrents behave when applied to various base propellants. The deterred propellants used in small arms and anti-aircraft guns were developed by a cut-and-try technique, and the sort of data required for rational design of large caliber weapons systems, e.g. burning rates in the deterred region, is extremely scarce. Since we lack an appropriate data base we have used thermochemical calculations and burning rate estimates to identify promising deterrent/base propellant combinations. The burning rate estimates were based on empirical fits to available closed bomb and strand burner tests with small arms propellants. Several promising deterrent systems are presented, including five candidates which exhibit a decrease in burning rate with no decrease in propellant energy. On the basis of these theoretical predictions, an experimental deterrent coating and analysis study has been initiated at Radford Army Ammunition Plant.

## INTRODUCTION

The requirement for increasingly higher muzzle velocities in large caliber weapon systems (LCWS) firing kinetic energy (KE) ammunition has led to theoretical and experimental investigations of high performance propulsion systems employing advanced technologies<sup>1</sup>. The goal of the effort has been to provide a significant increase in the muzzle velocities of existing weapon systems without changing the basic system hardware or operating limits. One highly successful and proven small arms technique for increasing performance under these constraints is that of deterred propellant. The task of applying this small arms deterrent technology to LCWS is complicated by the fact that the bulk of this technology has been empirically derived to meet the requirements of specific small arms applications with little direct emphasis on elucidating the fundamentals of deterrent technology<sup>2</sup>. This Edisonian development approach has been satisfactory in the past for small arms systems due to the relatively small amounts of propellant required and the simplicity of small arms

ballistic testing. It is cost prohibitive in LCWS, where large amounts of propellant are required and the cost of firing even a few rounds can exceed the costs of an entire small arms firing program.

The Ballistic Research Laboratory (BRL), US Army Armament Research and Development Command (ARRADCOM), has initiated a program to elucidate deterrent technology fundamentals so that high performance, deterred propulsion system may be efficiently designed and developed for any large caliber application. This program is currently using the M68, 105-mm tank gun firing KE ammunition as a convenient design base and test vehicle.

#### BACKGROUND

Producing a higher velocity for an existing projectile requires that more energy be imparted to the projectile during the ballistic cycle. The kinetic energy of a projectile at the end of the ballistic cycle, i.e., at muzzle exit, is proportional to the integral of the projectile base pressure-bore travel curve less resistive forces. One can increase the pressure integral while maintaining the allowable peak pressure by raising the trailing portion of the pressure-travel curve.

Raising the final portion of the pressure curve by increasing the mass rate of gas generation with the fraction of propellant burned is called progressive burning. Two factors that control progressivity are propellant surface area and linear burning rate (or rate of surface regression). These can be tailored to give the desired performance. At any given pressure, the amount of gas generated will be a function of the burning surface area. By selecting a grain geometry that gives an increase in surface as burning progresses, one can increase the mass burning rate and associated pressure after maximum pressure.

The second tailorable factor that controls mass burning rate is the linear burning rate. This is the rate at which the burning propellant surface recedes parallel to itself. In the interior ballistic cycle, the higher the burning rate, the more gas produced per unit time. By tailoring those portions of the propelling charge that burn during the final portions of the pressure curve to burn at a higher rate, more propellant can be consumed in the same time period and more gas will be generated, again, increasing progressivity. One method for providing such a differential burning rate propellant is through the use of deterrents; this has proven highly successful in small arms.

In order to apply this technology in a logical manner to LCWS, we must have a basic understanding of how deterrents work. A deterred propellant is one where the base grain formulation has had a chemical, the deterrent, impregnated into some or all of the exterior surfaces of the propellant granulation. This produces propellant grains with a non-uniform distribution of constituents consisting of a deterred surface layer surrounding or encapsulating the interior, original base grain regions (see Figure 1). The deterrent provides increased progressivity by lowering the base grain formulation's burning rate in the deterred layer. This is illustrated in Figure 2, which shows the closed chamber

burning rate versus pressure behavior for a standard M30, 7 perforation propellant with ethyl centralite (EC). Note that the initial portion of the propellant has a lower burning rate due to the deterrent. As the deterred layer is consumed, the burning rate returns to the standard M30 burning rate.

The deterred layer burns during the initial stages of the ballistic cycle, and it provides a slower rate of gas generation during these initial stages than would the undeterred base grain. The deterrent delays the development of the peak chamber pressure in time, and it allows the projectile to travel slightly farther down the bore. This increased projectile travel provides a large volume into which the combustion gases may expand. The result is a peak pressure lower than that produced by the same charge loading of the undeterred base grain formulation with identical geometry. This larger volume for pressure development allows the use of higher propellant loading densities in the chamber without exceeding the maximum allowable pressure. It also slightly raises the pressure in the trailing portion of the pressure curve as seen in Figure 3, thus increasing the integral and imparting greater energy and velocity to the projectile.

Since the reduced burning rate effect of the deterrent provides the desired increased progressivity, the obvious question is, how much improvement in muzzle velocity will deterred propellants provide? An initial technology demonstration effort by BRL in the 105mm, M68 tank gun firing KE ammunition and using deterred, 7-perforation M30 propellant produced a two percent increase in muzzle velocity over the standard M30 propellant at the same peak pressure<sup>3</sup>. This is not an insignificant gain as it translates into a 500-meter increase in effective range for this weapon system. Furthermore, theoretical studies as part of this initial effort have forecast that increases in velocity of as much as 10 percent are possible if the deterrent concept is combined with other proven propellant technologies, such as higher energy propellant formulations and more progressive grain geometry.<sup>3</sup> The promise of these performance gains has led to the present theoretical and experimental investigations. The goal of the present effort is to develop a rational, efficient, and expeditious method of optimizing the use of deterrent materials in a given base grain formulation.

#### CRITERIA FOR DETERRENT DESIGN AND SELECTION

Our first step in developing design criteria was to examine typical deterrent materials used in small arms development programs. Table I contains a listing of such materials. An initial analysis revealed that the one common characteristic of these materials is that they have negative (or only slightly positive) heats of formation. This is illustrated in Table II, which lists the theoretical adiabatic isochoric flame temperature (Tcv) and impetus of these materials at two concentrations in a homogeneous mixture with M9 propellant. Note that in all cases, that both the Tcv and impetus of M9 have been reduced.

Table I. Detergent Candidates

| Detergent           | Spherical Formula |      |       |   | $\Delta H_f$<br>Cal/mole | Name/Description        |
|---------------------|-------------------|------|-------|---|--------------------------|-------------------------|
|                     | C                 | H    | O     | N |                          |                         |
| DNT                 | 7                 | 6    | 4     | 2 | -14,498                  | Dinitrotoluene          |
| Camphor             | 13                | 16   | 1     |   | -78,100                  |                         |
| $\alpha$ -Cellulose | 6                 | 10   | 5     |   | -228,000                 |                         |
| N-Nap               | 10                | —    | 2     | 1 | -10,000                  | 1-Nitro Naphthalene     |
| S-51                | 5.143             | 9.73 | 1.638 |   | -100,096                 | Polyester Polymer       |
| EDM                 | 10                | 14   | 8     |   | -164,170                 | Ethylene Dimethacrylate |
| DOP                 | 10                | 14   | 4     |   | -115,992                 | Diphenyl Phthalate      |
| PBEN                | 13                | 19   | 1     |   | -57,702                  | Phenyl Benzoate         |
| TPU                 | 15                | 15   | 2     | 1 | -64,600                  | Diphenyl Urethane       |
| AKARI               | 13                | 12   | 1     | 2 | -28,300                  | Diphenyl Urea           |
| AKARII              | 14                | 14   | 1     | 2 | -34,100                  | Methyl Diphenyl Urea    |
| DBP                 | 10                | 22   | 4     |   | -203,900                 | Dibutyl Phthalate       |
| MC                  | 15                | 16   | 1     | 2 | -14,180                  | Dimethyl Diphenyl Urea  |
| Vinsol              | 16                | 15   | 3     |   | -143,900                 | Trade Name              |
| AKARIII             | 15                | 16   | 1     | 2 | -35,200                  | Ethyl Diphenyl Urea     |
| EC                  | 17                | 20   | 1     | 2 | -17,670                  | Diethyl Diphenyl Urea   |
| Polystyrene         | 9                 | 8    |       |   | -17,422                  |                         |
| COS                 | 16                | 30   | 4     |   | -344,210                 | Diethyl Sebacate        |
| Irraton             | 7                 | 12   |       |   | -22,020                  | LOVA Binder Polymer     |
| Polyethylene        | 2                 | 4    |       |   | -12,700                  |                         |
| Carbowax 400        | 10                | 16   | 2     | 2 | -181,000                 | Carbowax 400            |

Table II. Detergents Ranked by  $10^3 T_{cv}$  at the Same Concentrations

| Base Gran.          |      | $T_{cv}$   | Impetus       | $T_{cv}$  | Impetus       |
|---------------------|------|------------|---------------|-----------|---------------|
|                     |      | (K)        | (J/g- $m^2$ ) | (K)       | (J/g- $m^2$ ) |
| <u>Detergent</u>    |      |            |               |           |               |
|                     |      | 1.5% Conc. |               | 15% Conc. |               |
| DNT                 | 3664 | 1165       |               | 3440      | 1147          |
| Camphor             | 3565 | 1117       |               | 3221      | 1040          |
| $\alpha$ -Cellulose | 3684 | 1121       |               | 3959      | 1042          |
| N-Nap               | 3452 | 1138       |               | 2939      | 1055          |
| S-51                | 3396 | 1076       |               | 2814      | 931           |
| EDM                 | 3375 | 1122       |               | 2820      | 1022          |
| DOP                 | 3343 | 1107       |               | 3723      | 983           |
| PBEN                | 3299 | 1097       |               | 2636      | 972           |
| TPU                 | 3285 | 1111       |               | 2519      | 978           |
| AKARI               | 3284 | 1109       |               | 2612      | 972           |
| AKARIII             | 3270 | 1111       |               | 2590      | 973           |
| DBP                 | 3269 | 1129       |               | 2605      | 972           |
| MC                  | 3265 | 1114       |               | 2582      | 978           |
| Vinsol              | 3256 | 1101       |               | 2567      | 963           |
| AKARIII             | 3250 | 1108       |               | 2553      | 967           |
| EC                  | 1236 | 1110       |               | 2530      | 972           |
| Polystyrene         | 3191 | 1088       |               | 2407      | 906           |
| COS                 | 3110 | 1098       |               | 2334      | 935           |
| Irraton             | 3022 | 1096       |               | 2220      | 916           |
| Polyethylene        | 2961 | 1087       |               | 2150      | 997           |
| Carbowax 400        | 2767 | 995        |               | 1640      | 712           |

A number of factors must be known to design a deterred propellant system with a minimum of trial and error<sup>2</sup>. These factors include chemical compatibility, chemical stability, deterrent penetration, ignitability, and burning rate modification. The most basic requirement is that the deterrent material be chemically compatible with the base grain ingredients. This is usually determined for short term purposes by such compatibility tests as Vacuum Stability, the Heat (methyl violet) test, thermogravimetric analysis, Taliani, and differential scanning calorimetry. The long-term chemical compatibility is generally determined by accelerated aging studies where the ingredients are mixed together and stored at elevated temperatures for long time periods. Such factors as weight loss, color changes, changes in mechanical properties, and changes in chemical species concentrations are usually monitored. As a general rule the more reliable data are obtained at temperatures around 303-313 Kelvin over long time periods<sup>5</sup>.

The next basic requirement is that the deterrent material be capable of penetrating the base grain matrix. Penetration is controlled by a large number of factors: the solubility of the deterrent in the coating solvent system; solvent levels in the base grain; nature of the base grain binder and plasticizer system; coating temperature; degree of agitation; deterrent melting point; deterrent molecular size; coating time; chemical interaction of functional groups on the deterrent and base grain ingredients; and deterrent concentration in the coating solution.

Most detergents are applied at elevated temperatures in a solvent solution or suspension to propellant grains kept agitated by some means. The agitation prevents clumping of the grains during the impregnation, or as it is commonly called, the coating process, and encourages a uniform distribution of deterrent into all the surfaces of the grains, including the perforations. The result is illustrated in figure 1a.

In another experimental method the propellant strands are dipped into a super-saturated deterrent solution after extrusion but prior to cutting the strands into grains. This process allows the deterrent to penetrate only into the outer surface of the strand and not into the perforations or grain ends, and is illustrated in Figure 1b.

A similar coating of the extruded strands with a solution that does not penetrate into the surface produces a coating exterior to the strand surface. Materials applied in this manner are generally referred to as inhibitors and are illustrated in Figure 1b. Inhibitors prevent burning of the exterior wall of the strands or grains during some or all of the gun combustion cycle. As a rule, deterrent candidate materials that do not penetrate into the base grain, but form an inhibitor-like surface coating, cause clumping of the grains during storage and are unacceptable.

It is no trivial matter to determine the proper processing conditions and materials to produce an acceptable deterrent coating. The penetration is usually determined by chemical analysis of the bulk amount of deterrent impregnated into the grain, followed by closed base and ballistic firings<sup>12</sup>. Penetration depth can also be determined by such methods as optical analysis of thin sections of propellant grains where the deterrent

changes the optical density of the deterred region (sometimes heightened by staining techniques)<sup>2</sup>. Other methods can include chemical analysis of thin sections of the deterred regions to determine penetration depth and actual deterrent concentration in this layer<sup>6</sup>.

Once the required penetration has been achieved, as determined by successful ballistic performance, the deterrent material should not migrate with time and temperature within the grain. Most deterrents are believed to penetrate the grain via a diffusion with binder interaction mechanism which results in a deterrent concentration profile as seen in Figure 4<sup>6</sup>. Note that the deterrent concentration is relatively constant throughout the deterred region with a very steep gradient at the deterrent penetration front. This produces a propellant grain which can easily be thought of as a two layered system consisting of the surface, low burning rate deterred layer surrounding the internal, higher burning rate, base grain regions. Experimental evidence with nitrocellulose (NC) binder systems suggests that the formation of, and the stability of, this type of deterrent concentration profile is the result of hydrogen bonding between the deterrent molecule and the binder<sup>2,6,10</sup>. This type of concentration profile may not hold true for polymeric deterrents which contain few chemical functional groups available for hydrogen bonding but rather depend on the overall size of the molecule to prevent migration.

Deterrent migration is generally evaluated by closed bomb determination of relative quickness (RQ) and relative force (RF). The freshly deterred propellant is tested and is then stored at some constant, elevated temperature (30-50°C) for at least three months. The RQ and RF are then determined again. Changes in RQ are primarily attributed to deterrent migration<sup>12</sup>.

Deterrents tend to increase the ignition delay. In rapid firing weapons this can be a significant factor. In LCWS, with relatively slow rates of fire, this is not a major effect and should easily be resolved through a stronger ignition system. Ideally, the deterrent with the shortest ignition delay should be chosen when all other factors remain the same.

The base grain burning rate modification by the deterrent is the final and most important criterion. The deterrent may be chemically compatible, penetrate into the base grain, have good ignition, and not migrate, but all of these features are irrelevant if the deterrent does not reduce the burning rate of the base grain to produce the desired ballistic results. Only deterrents that can adequately modify the base grain burning rate in reasonable concentrations need be considered for compatibility, penetration, and migration evaluation with a given base grain formulation. The degree of this burning rate decrease effect would provide an initial screening criterion for deterrent/base grain selection and would greatly simplify and expedite the development process; it would also serve as input data in interior ballistic computer simulation codes to guide the development program.

## MODELING OF DETERRED SYSTEMS

Modern interior ballistic codes do a good job of modeling the performances of undeterred charges. The additional information required by these codes to model deterred systems consists of the thermodynamic data for the deterred region, the depth of deterrent penetration into the web, and the burning rate of the deterred region.

The required thermodynamic data can be easily calculated and consists of the impetus (force),  $T_{cv}$ , specific heats ratio of the combustion gases ( $\gamma$ ), the covolume, and for some codes, the average molecular weight of the combustion gases. The concentration of the deterrent in the deterred layer must be known to make these calculations. Since deterrent is confined to the outer layer, the actual deterrent concentration in the deterred layer is greater than the bulk concentration. While unique for each base grain, deterrent and weapon system application, the bulk concentrations of small arms deterrents typically range from one to ten percent of the total propellant weight with concentrations of the deterrent in the deterred region ranging approximately from ten to thirty percent<sup>7</sup>.

Figures 5 and 6 show the changes in impetus and  $T_{cv}$  as a function of deterrent concentration in an M9 propellant base grain. As seen in these figures, the  $T_{cv}$  and impetus of the deterred layer in the 10 to 30 percent range is considerably different from the no-deterrent (base grain formulation) case. This supports our model of a deterred grain as, in the simplest case, consisting of two different propellants.

The burning rate of the deterred layer is the next piece of information required. For each base grain formulation, this will depend on the deterrent, its concentration, and the concentration profile in the deterred layer. With the change in base grain burning rate as a function of deterrent concentration known, one could easily simulate existing deterred systems. Conversely, this data could be used to design deterred propulsion systems. The problem is that there is very little deterred layer burning rate data available. This situation must be rectified before effective modeling can occur.

Since burning rate reduction is the prime effect of deterrents, let us put aside the thermochemical effects for the moment. By considering a deterred grain as a two-propellant system, based on the concentration profile of Figure 4, interior ballistic codes could be used to determine the optimum differences in burning rates and the thickness of the deterred layer (depth of penetration) to provide the desired ballistic effect. From these results, a deterrent candidate could be selected that will produce the required reduction in burn rate as a function of concentration and the depth this concentration must penetrate to give the required results. This would not only allow efficient selection of deterrents, but also serve as a design guide in producing the required propellant.

### DETERRED REGION BURNING RATE DATA

While there is a wide variety of available deterrent materials and an ever-increasing number of energetic base grain formulations, there

has been little direct experimental burning rate determination of compositions typical of deterred regions. However, there are at least two studies where the burning rate of a homogenous deterred system has been determined. In these studies various levels of deterrent were incorporated as part of the propellant composition during manufacture. This resulted in grains or strands with a uniform distribution of constituents throughout the grain or strand. Table III contains a summary of the compositions and reduced burning rate data of these studies.

In a study performed for BRL by Reifler and Lowery of the Olin Corporation, two sets of propellants were produced to simulate single and double base compositions. The double base compositions contained 8-10 percent nitroglycerine (NG). Both compositions used NC with approximately 13.1% nitrogen content. Duplicate ball propellant lots at various dibutyl phthalate (DBP) levels (a maximum of 16 percent DBP) were produced and the closed bomb burning rates determined. The initial effort was to determine the burning rate as a function of NG and DBP concentrations. Stiefel re-analyzed the data and correlated the burning rate as a function of the  $T_{cv}$  of each composition<sup>7</sup>. The correlation of the burning rate coefficient with a constant exponent of 0.8053 was determined to be

$$b = -0.8340 + 8.3956 \times 10^{-4} T_{cv}$$

where

$$r = bp^n$$

and  $r$  = mm/sec burn rate,  $b$  = coefficient,  $P$  = MPa,  $n$  = 0.8053.

Stiefel's plot of the burning rate coefficient versus  $T_{cv}$  with the Riefler and Lowery data is given in Figure 7, and shows the best fitting straight line.

In a similar, unpublished effort, Moy determined the strand burner burning rate of a homogenous series of propellants with ethyl centralite (EC) deterrent. The base grain composition consisted of NC (12.6 percent nitrogen) with 2 percent 2dinitro diphenyl amine stabilizer. Propellant formulations with EC concentrations of 0, 1, 5, 10 and 20 percent were produced and the strand burning rates determined. Our evaluation of these data at BRL determined the burning rate as a function of EC concentration to be

$$r_d = (0.96 - 0.12 \ln [EC]) r_0$$

where  $r_d$  = mm/sec burning rate of EC containing composition,  $r_0$  = mm/sec burning rate of composition without EC, and  $[EC]$  = the EC concentration in percent.

The constant coefficient correlation of the Moy data between the exponent and  $T_{cv}$ , with a coefficient of correlation ( $R^2$ ) of 0.995 is

$$n = 0.82945 + 5.5 \times 10^{-5} T_{cv}$$

Table III - Summary Homogeneous Detergent Burn Rate Data

3. RIEFFER AND LOWERY CLOSED BOMB BALL PROPELLANT LINEAR BURN RATE STUDY<sup>8</sup>

| PROP  | V. DOP              | LINEAR BURNING MODEL |                       | BLAKE TCV<br>(E) |
|---|---------------------|----------------------|-----------------------|------------------|
|   |                     | $5 \times 10^{-4}$   | $\eta$                |                  |
| <b>SINGLE BASE: NC (13.1% N)</b>  |                     |                      |                       |                  |
| 1.34  | 1.34                | 13.750               | 0.7988                | 2961             |
| 1.59  | 1.59                | 14.491               | 0.7665                | 2459             |
| 1.74  | 1.74                | 15.309               | 0.7440                | 2245             |
| 2.11  | 16.26               | 17.183               | 0.7206                | 2064             |
| <b>DOUBLE BASE: NC (13.1% N)</b>  |                     |                      |                       |                  |
| 12.59   | 0.99                | 10.145               | 0.8321                | 3155             |
| 9.75  | 1.00                | 9.137                | 0.8301                | 2843             |
| 9.10  | 13.30               | 11.932               | 0.7424                | 2199             |
| 9.10  | 16.65               | 12.509               | 0.7489                | 2034             |
| <b>B. B. A. HIGH STRAND BURNER DETERGENT STUDY<sup>9</sup> - 160-320 MPa DATA</b> |                     |                      |                       |                  |
| PROPELLANT  | <u>BASE FORMULA</u> |                      | <u>DETERGENT</u>      |                  |
|   | 93.91 NC (13.1% N)  |                      | 100% ETHYL CENTRALITE |                  |
|   | 0.2% 2NQPA          |                      |                       |                  |
| STRAND  | V. FC               | $5 \times 10^{-4}$   | $\eta$                | BLAKE TCV<br>(E) |
|   |                     |                      |                       |                  |
| 3.0   | 1.986               | 1.020                | 3086                  |                  |
| 1.0   | 1.62                | 1.3648               | 2990                  |                  |
| 1.0   | 1.000               | 1.9993               | 2647                  |                  |
| 10.0  | 1.90                | 0.9807               | 2331                  |                  |
| 20.0  | 0.70                | 0.2157               | 1879                  |                  |

where

$$r = bP^n$$

and,  $r$  = mm/sec,  $b$  = 0.003 and  $P$  = psi.

The constant exponent correlation between coefficient and  $T_{cv}$  with an  $R^2$  of 0.995 is

$$b = (4 \times 10^{-5} + 6.056 \times 10^{-8} T_{cv})^{25.4}$$

where

$$r = bP^n$$

and

$$r = \text{mm/sec, } n = 0.9776 \text{ and } P = \text{psi}$$

A plot of the correlation of both constant exponent and constant coefficient with composition  $T_{cv}$  is seen in Figure 8. The best fitting straight lines are shown.

We attempted to correlate the burning rate effects of both these studies into a single expression. Numerous combinations of  $T_{cv}$ , average gas molecular weight, covolume, and adiabatic pressure with the reported burning rates from a first to a third degree multivariable linear regression analysis failed to give a useful correlation. Reasons for this lack of correlation could include the differences in burning rate determination methods and the differences in nitrogen content of the NC (13.1 and 12.6 percent). However, in both studies the burning rates are strong linear function of the composition  $T_{cv}$ , which ranged from 1900 to 3200 Kelvin in both efforts. More experimental data are needed to develop a unified expression for predicting deterred region burning rates in any base grain over a wide  $T_{cv}$  range.

#### DESIGN OF DETERRED SYSTEM

Deas and Trafton used the Riefler and Lowery burning rate study formula (based on ingredient concentrations) to determine the theoretical optimum performance obtainable with a hypothetical high-energy double-base propellant, roughly equivalent to M9 propellant, with DBP deterrent<sup>3</sup>. The design test bed was the M68, 105mm tank gun firing KE ammunition. Double-base propellant was chosen, as it is an energetic propellant that is readily producable. The goal was to predict performance gains available from a propellant (with a higher energy than the standard M30) both with and without a deterred layer.

For their analysis, the bulk concentration was set at 2, 4, and 6 percent DBP. At each bulk concentration the deterred surface layer concentration (SLC) was varied at 7, 9, 13, and 17 percent DBP. The depth of penetration was determined by the SLC. Thus the higher the SLC, the less the deterrent could penetrate. The web at each charge weight and SLC was varied to give the same peak pressure. The results are seen in Figure 9.

This theoretical analysis determined the optimum conditions to be a 2-4 percent bulk concentration with a 7-9 percent SLC. This deterrent combined with a 19-perforated base grain gave a predicted 10-percent velocity improvement over the standard M30 propellant in the 105mm design system. Three percent of this gain was due to the deterrent.

DBP was chosen as the deterrent for this theoretical study because of the availability of burning rate data but it is not a realistic choice. DBP tends to migrate in propellants containing more than 20 percent NC<sup>2</sup>; and thus a problem remains of selecting a suitable replacement for DBP in order to test these predictions experimentally.

We used the strong dependency of the deterred layer burning rate on Tcv to develop a useful design tool for selecting a DBP replacement. We assumed that any chemical effects of deterrent candidates on the base grain burning rate are minimal and accounted for in the calculation of Tcv. As MB is a double base propellant, Stiefel's correlation was used as a yardstick to compare the relative burning rate effects of candidates to that of DBP. Deterrents that have Tcv's similar to DBP at the same concentration should modify the base grain burning rate almost the same as DBP. Thus viable, direct DBP replacements could be selected in a rational manner. Then other factors such as penetration behavior and migration effects could be used for evaluation and final selection.

One note of caution is seen in Figure 6. Note that all the curves are more or less linear down to about 2200 K. Below this temperature we recognize that our linear correlation model will be invalid and care should be exercised in making comparisons.

Table II lists a number of candidates to replace DBP which are ranked by TCV at a 7.5 percent concentration in MB. Based on the Stiefel correlation, a +70 K Tcv variation around the Tcv of DBP at 7.5 percent gives a burning rate variation of about  $\pm 5$  percent. This provides a reasonable rule of thumb to select a direct DBP replacement. There are a number of materials in the table that fall within the range.

The candidates in this range are all close to the same molecular weight but should have different penetration and migration effects. Diphenyl phthalate (DPP), a slightly higher molecular weight analog of DBP, has been used to replace DBP in highly plasticized binders (>20 percent NC). Due to the larger phenyl group, DPP could be expected to penetrate less than DBP, but also to migrate less than DBP. Broadman and Devine also suggest this type of behavior by DPP is due to the greater electron contributing effect of the phenyl group which makes the carbonyl oxygen of the phthalate group more electronegative, thus forming stronger hydrogen bonds with NC than DBP<sup>10,11</sup>. The other materials in the group around DBP should demonstrate similar subtle penetration and migration differences.

Deterrents with higher Tcv's than DBP will require higher concentrations to achieve the same burning rate modification. Camphor at the 15 percent concentration has about the same Tcv as DBP at 7.5 percent. Thus, for the same burn rate change, camphor would need about twice the concentration as DBP at the same penetration. The opposite trend should be true for

such materials as Arcwax, which is about 15 percent cooler than DBP at the 7.5 percent concentration.

Once the deterrents are ranked by  $T_{cv}$ , the next factors to consider are the gross penetration and migration effects. Due to the high plastisizer content of M9, molecules similar in size to DBP may migrate. One solution is to use much larger or even polymeric molecules. Arcwax is such a large molecule, but due to its size, there may be a problem in achieving the same penetration depth at the lower concentration necessary to give the same burning rate effect as DBP. Due to size alone, Arcwax will form a narrow, high SLC with very low burning rates. GS4-Paraplex (Herkote) is a polyester polymer and should show similar size effects as Arcwax but, with its higher  $T_{cv}$  than DBP at 7.5 percent, the burning rate modification of GS4 will be less than Arcwax at the same SLC. Thus GS4, while still tending to form a high SLC, is favored over Arcwax. Another advantage of GS4 is that, along with its high molecular weight, it has sites available for hydrogen bonding. Both of these are features which prevent migration.

Of direct interest to producing a propellant with any deterrent is the sensitivity of ballistic performance to SLC variation, i.e., depth of penetration. The general trend seen in Figure 9 is, that the lower the bulk concentration, the less sensitive the performance is to depth of penetration, i.e., the SLC curves are closer together. This lowered sensitivity provides an important design feature. Using low bulk concentrations will allow greater flexibility in the processing conditions which control depth of penetration. This flexibility will help insure lot-to-lot performance uniformity, a major problem in deterred propellant manufacture.

Should it become a problem, the deeper penetration required by the four percent bulk concentration could be achieved by harsher deterring conditions such as stronger solvent systems, higher temperatures and more residual solvent in the base grain. These conditions in a double base system such as M9 tend to leach out the NG. This reduces the overall energy of the finished propellant and the NG contamination of the coating solvent system complicates processing. The two percent bulk concentration plot shows far less sensitivity to depth of penetration. This deterrent level should make production easier and still provide significant performance gains. While this analysis indicates trends, experimental penetration and migration data with M9 is needed before a rational choice can be made.

#### EXPERIMENTAL EFFORTS

A development program is underway at Radford Army Ammunition Plant (RAAP) to evaluate those concepts for demonstration in the 105-mm tank gun. The initial stage of this effort consists of deterring M9 grains in the laboratory under identical processing conditions (typical of large scale production) at two or more coating levels. The initial selection of deterrent candidates will come from Table II. The change in base grain RF and "Q will be determined as well as the bulk and deterred layer concentrations. Determination of the deterred layer burning rate will be attempted via closed bomb analysis.

Based on the penetration and burning rate data, several large scale lots of ballistic and homogeneous deterred lots will be produced. The ballistic lots will be deterred and, after analysis at RAAP, test fired at BRL. The homogeneous lots will be produced for closed bomb burning rate analysis at BRL. This effort will not only develop a high energy, deterred propellant for the M68 tank gun, but also validate the design concepts presented.

The initial laboratory results from RAAP indicated that the ethyl alcohol-water solvent system typically used at RAAP to soften small arms grains prior to adding the deterrent was having two main effects. The first was that the M9 grains were being softened too much, causing clumping and grain deformation upon drying. An explanation is that the high NG content in M9 may be causing the grains to be too soluble in the normal softening solution. Another factor may be that propellant grains for the 105-mm gun are much larger than a typical small arms grain. This means that the LCWS grains have a much smaller surface area per unit mass than small arms grains. Thus it takes longer for the same amount of alcohol to penetrate into the smaller surface area of the same mass of LCWS grains. This results in a higher alcohol concentration for a longer time period at the LCWS surface. This coupled with the higher solubility of M9 in the ethyl alcohol softening agent may be producing the observed results. The concentration of the alcohol is being reduced to solve this problem.

The second effect of the softening solvent found in the laboratory study was that, even under the mild, small arms-type coating conditions, the NG was being leached out. Closed bomb analysis of samples run through the processing conditions without a deterrent showed a significant reduction in RQ. This effect is being analyzed further. Should these initial results prove valid, this may provide a useful method for generating an exterior, low burning rate region in propellants using high concentrations of energetic plasticizers.

The next step was to use the laboratory results to coat 22.5 kilogram (50 pounds) batches of M9 propellant in the production coating equipment with C-54 and methyl centralite. The first attempts were at a coating temperature of 67°C. This resulted in excessive softening of the grains with both deterrents. From this effort it was discovered that about 90 percent of the deterrent added in the coating solution was being taken up by the grains.

From these results it was decided to reduce the coating temperature. Subsequent coatings were made at 55 and 45°C with the same solutions and deterrent concentrations as attempted at 67°C. The coatings at both temperatures appeared acceptable. The analysis of these efforts are still in progress and further data was unavailable at the time this report was written.

#### FUTURE EFFORTS

In order to expand the deterrent technology data base, several future efforts are planned. While M9 is high in energy and easily produced, its high flame temperature will cause barrel erosion and possibly flash problems. Some of these effects will be ameliorated by the deterrent. Future efforts will include developing deterrents for

TABLE IV Energetic Detonants

| DETOXIN     | EMPIRICAL FORMULA |    |   |   | $\Delta H_f$<br>cal/mole | NAME/DESCRIPTION              |
|-------------|-------------------|----|---|---|--------------------------|-------------------------------|
|             | C                 | H  | O | N |                          |                               |
| PROPYL NENA | 5                 | 11 | 5 | 3 | + 40,603                 | N-3-Alkyl Analogs of          |
| METHYL NENA | 6                 | 13 | 5 | 3 | + 46,702                 | Nitrate ethyl                 |
| PENTYL NENA | 7                 | 15 | 5 | 3 | + 49,808                 | Nitramine (NENA)              |
| HEXYL NENA  | 8                 | 17 | 5 | 3 | + 52,915                 |                               |
| HEPTYL NENA | 9                 | 19 | 5 | 3 | + 56,224                 |                               |
| HEPTYL      | 9                 | 19 | 7 | 3 | + 57,100                 | 3,3-Bis (azidomethyl) -octane |
| NBN         | 5                 | 5  | 2 | 3 | + 37,062                 | 1-Nitroso Butyl Nitrate       |
| GAP         | 3                 | 6  | 2 | 3 | + 33,700                 | Glycidyl Urea Polymer         |

Table V. Energetic Detonants Rated by Tex in 10<sup>6</sup> Propellant

| DETOXIN     | 25% Conc |         | 15% Conc |         |
|-------------|----------|---------|----------|---------|
|             | Tex      | Impetus | Tex      | Impetus |
| IP          | 3713     | 1192    | 3564     | 1223    |
| BN          | 3707     | 1191    | 3545     | 1190    |
| PROPYL NENA | 3706     | 1184    | 3549     | 1187    |
| PENTYL NENA | 3464     | 1152    | 3435     | 1176    |
| WT          | 3608     | 1165    | 3440     | 1142    |
| NBN         | 3613     | 1163    | 3430     | 1173    |
| HEXYL NENA  | 3617     | 1179    | 3371     | 1169    |
| HEPTYL NENA | 3692     | 1177    | 3297     | 1159    |
| HEPTYL      | 3643     | 1178    | 3150     | 1157    |

high energy, lower flame temperatures propellants based on solid filled, nitramine systems. Similar efforts to the present M9 program are planned using this type of propellant. This will also include determining deterred homogeneous burning rates of various deterents with these nitramine base grain formulations.

Traditional deterents lower both the impetus and  $T_{cv}$  of the base grain. This reduction in energy offsets the gain of burning rate control by requiring an increased charge loading to replace the energy lost due to the deterrent. What if the impetus could be maintained while the flame temperature and burning rate is reduced? There are several low  $T_{cv}$ , high energy materials being developed to replace the high  $T_{cv}$  plasticizers such as NG and to replace NC as the only choice for a high energy propellant binder. Table IV is a listing of some of these materials and Table V is a comparative ranking of the  $T_{cv}$  at the same concentration in M9. A useful comparison can be made with DNT, the most energetic deterrent in common US propellants, and which can be seen to have a high  $T_{cv}$  (Table II). By assuming that the burning rate is a function of  $T_{cv}$ , the correlation of burning rate with flame temperature determined by Stiefel can be used to predict the change in burning rate produced by the materials.

Figure 10 is a comparison of the predicted change in M9 burning rate at a 20-percent concentration of a variety of standard and energetic deterrent materials as compared to the change the same concentration of DNT is predicted to give. With DNT as our yardstick, GAP, NBN and Propyl NENA do not lower the M9 burning rate as much as DNT. Indeed GAP and NBN are being developed to provide increased burning rates. The alkyl NENA's above Butyl NENA as well as the BAMO polymer show promise.

Figure 11 compares the impetus of these materials at a 20-percent concentration to the M9 base grain impetus. Note that a 20-percent concentration of Heptyl NENA has about the same impetus as the same concentration of DNT, while providing a greater burning rate reduction due to the lower  $T_{cv}$  than DNT as seen in Figure 10.

While admittedly a crude analysis, the data do suggest that the development of energetic, low flame temperature deterents is possible. These materials will be included in the laboratory scale coating efforts at RAAP in FY82.

#### CONCLUSIONS

Designing and developing deterred propellant systems for LCWS in a rational manner requires knowing the effect of a deterrent on the base grain formulation's burning rate. From an analysis of the limited deterred layer burning rate data available, a method for predicting this burning rate effect has been developed based on the strong linear correlation of burning rate with the adiabatic, isochoric flame temperature of the deterred region down to about 2000 K. This theoretical correlation based on thermochemistry does not predict exact burn rate effect in

every base grain, but does provide relative comparisions. Using this correlation several promising deterrent candidates for an M9 double base propellant have been selected for evaluation in the 105-mm tank gun. A significant achievement of this methodology is the selection of five deterrent candidates which exhibit a decrease in burning rate without the energy loss associated with traditional deterrents. This analysis is being used to guide a deterrent development program now ongoing at RAAP.

#### ACKNOWLEDGMENTS

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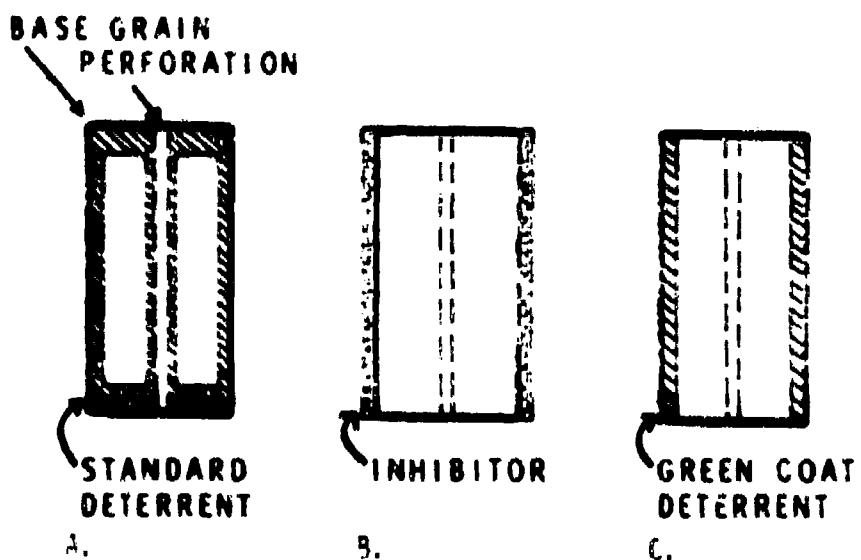


Figure 1. Examples of Deterrent- and Inhibitor-Treated Propellants

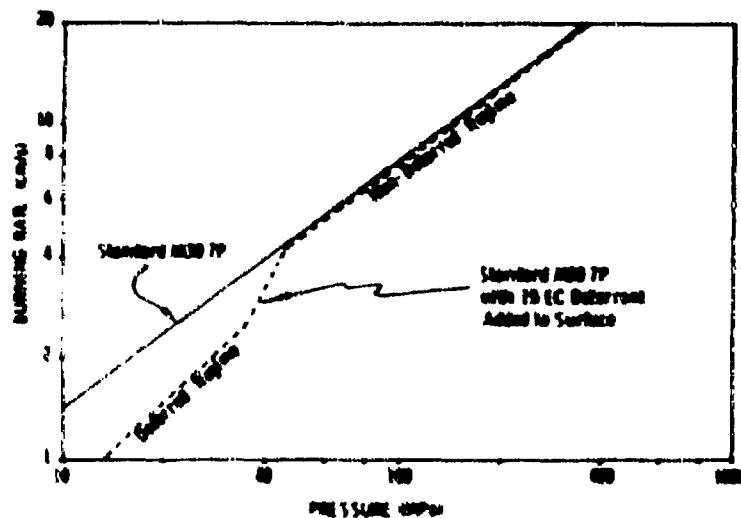


Figure 2. Burning Rate of M30 7P Propellant, Standard and Deterred (Reference 3)

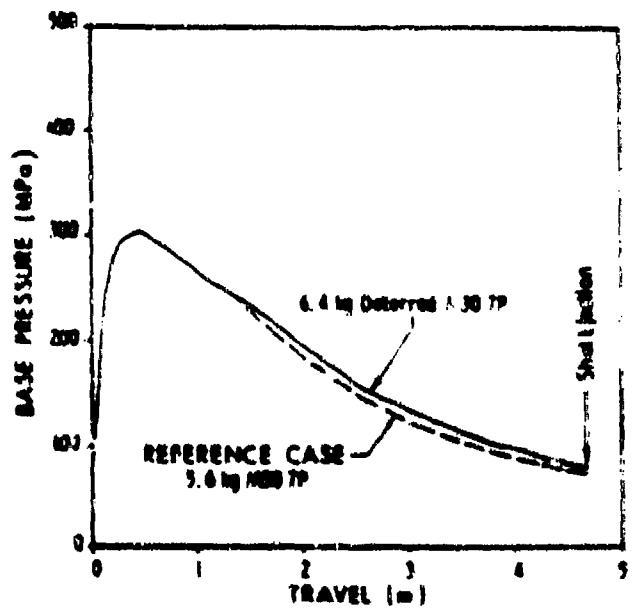


Figure 3. Base Pressure vs Travel with Standard and Deterrred M30 7P Propelling Charges (Reference 3)

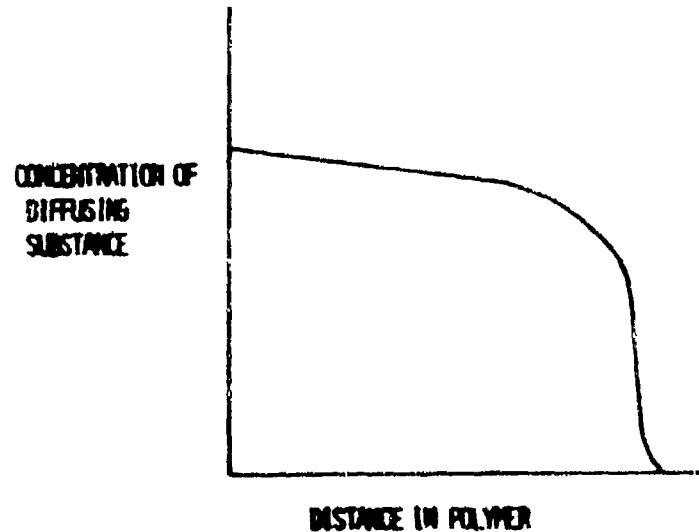


Figure 4. Deterrent Concentration Profile Produced by a Diffusion with Interaction Mechanism (Reference 6)

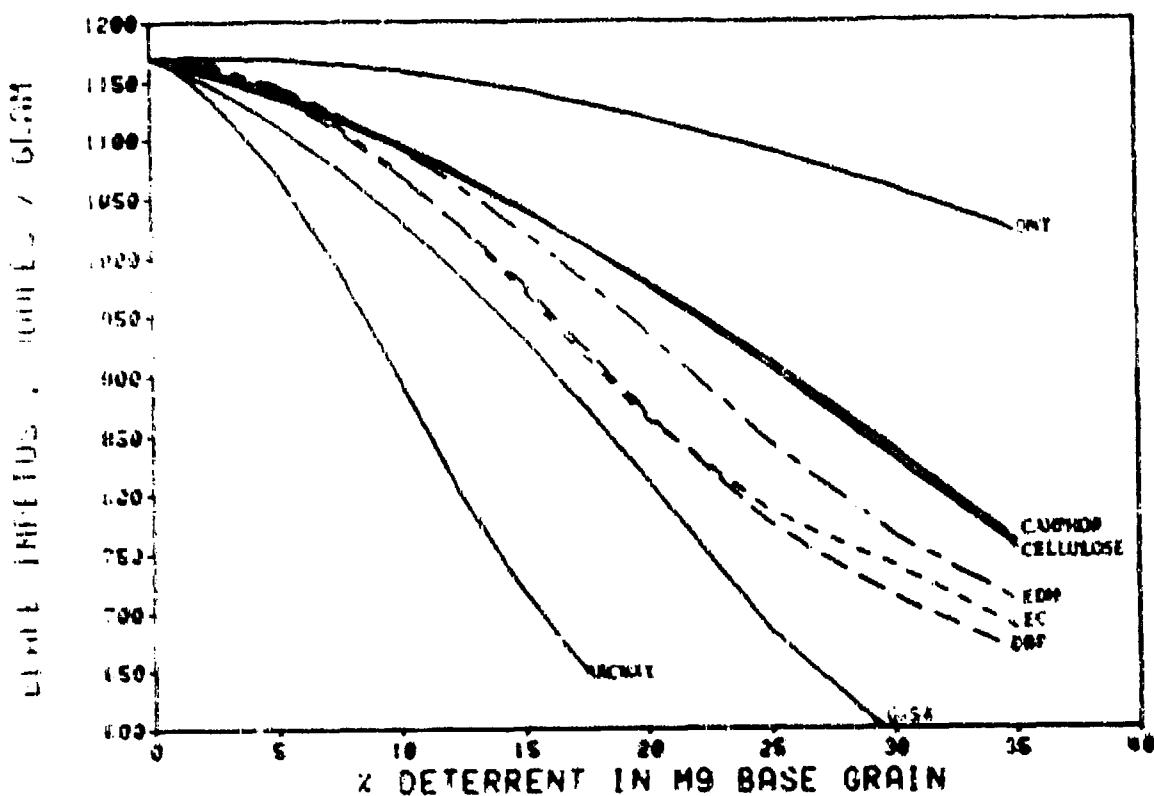


Figure 5. Plot of Change in Impetus of Selected Detergents from Table II in an M9 Base Grain.

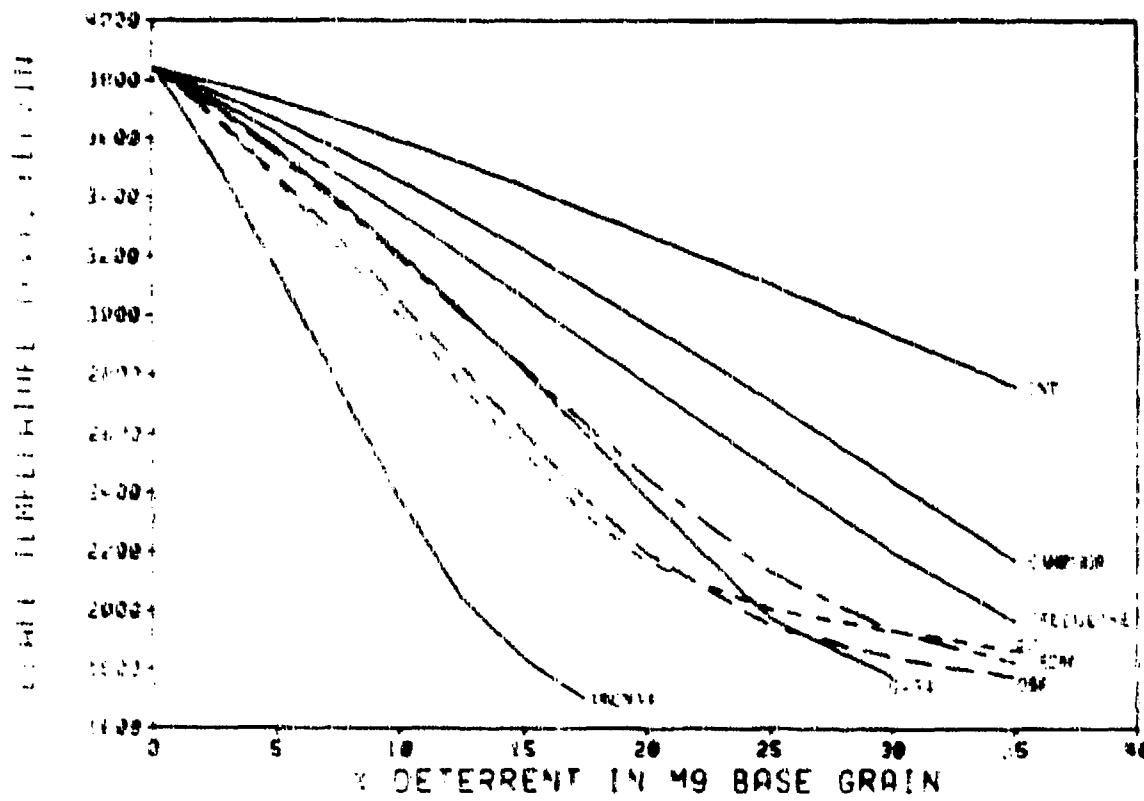


Figure 6. Plot of Change in Isochoric Adiabatic Flame Temperature of Selected Detergents from Table II in an M9 Base Grain.

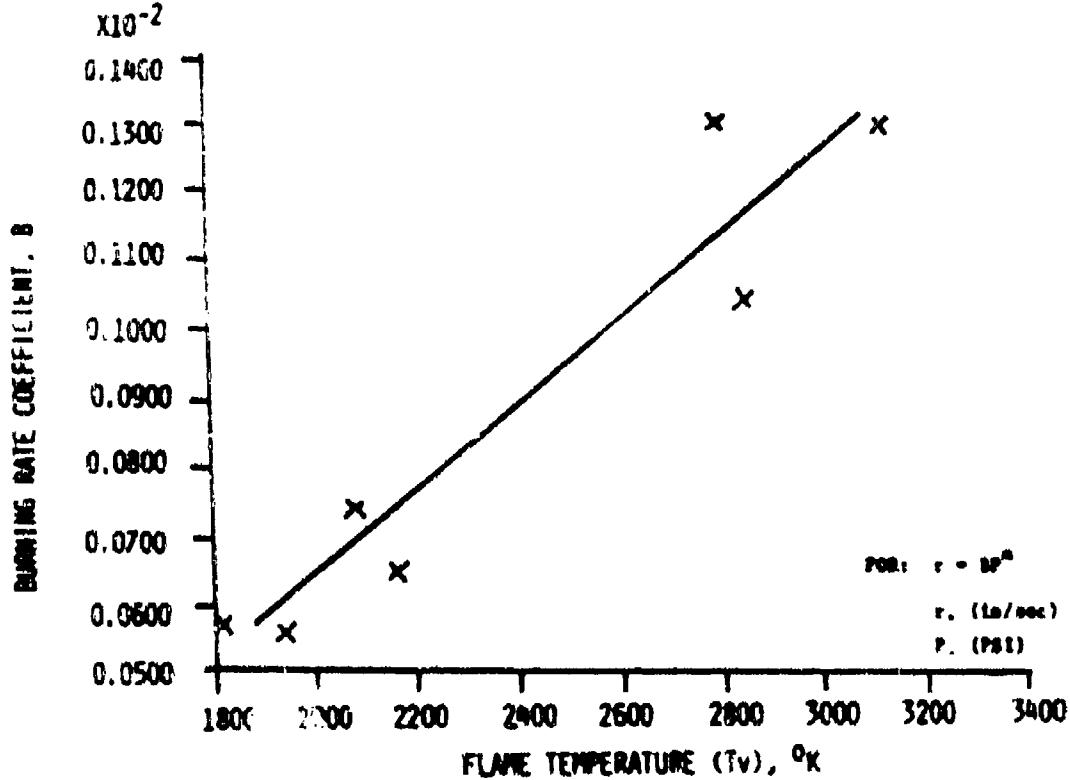


Figure 1. Correlation of Rieffler and Lowry Burning Rate Data, Ref. 13, with Flame Temperature by Stiefel, Ref. 1.

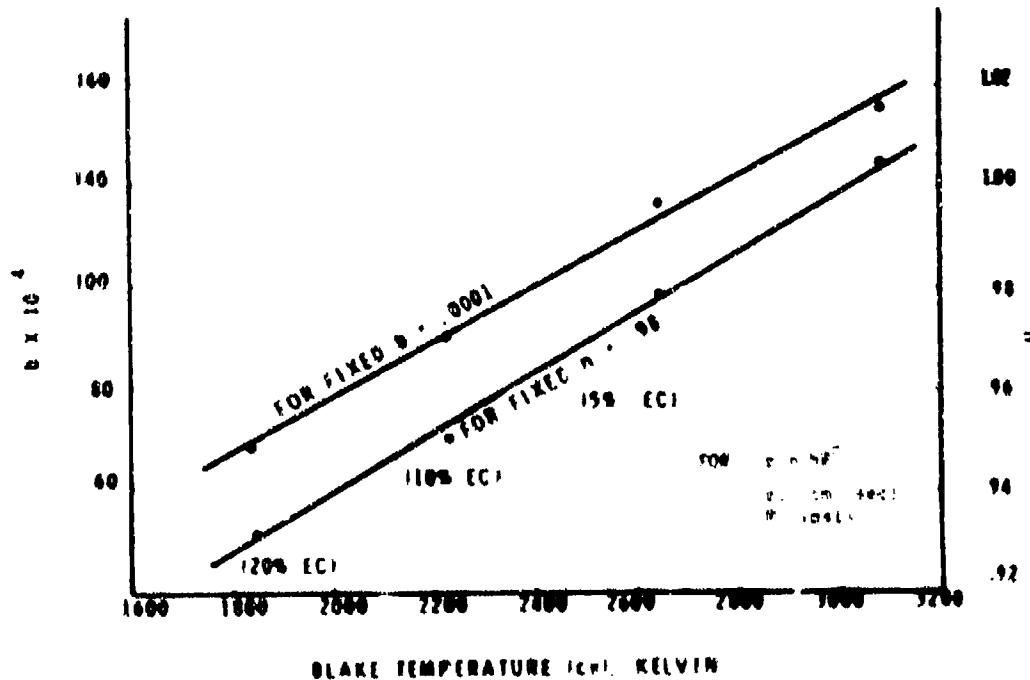


Figure 2. Correlation of von Kármán Burning Rate Data, Ref. 9, with Flame Temperature with Both a Constant Exponent,  $n$ , and Coefficient  $b$ .

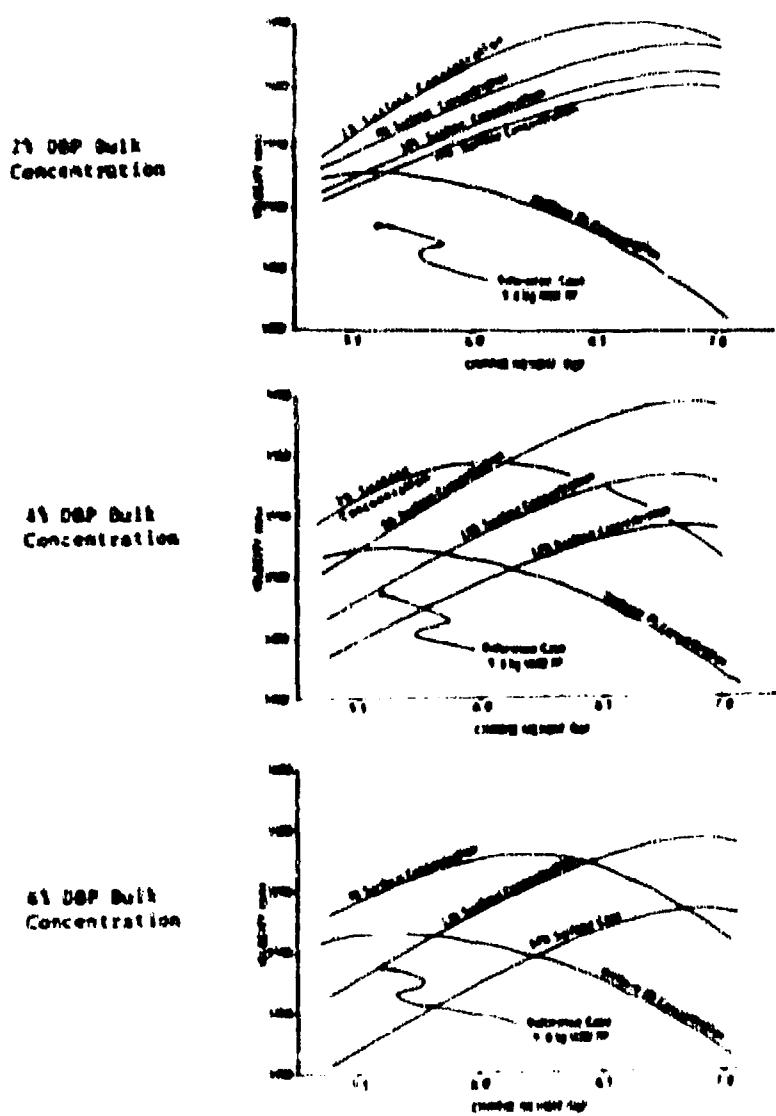


Figure 9. Predicted Velocity of Detonated High-Energy 19-Perforation Propellant with Varying Detergent Levels ( $P_{max} = 415$  MPa)  
(Reference 3)

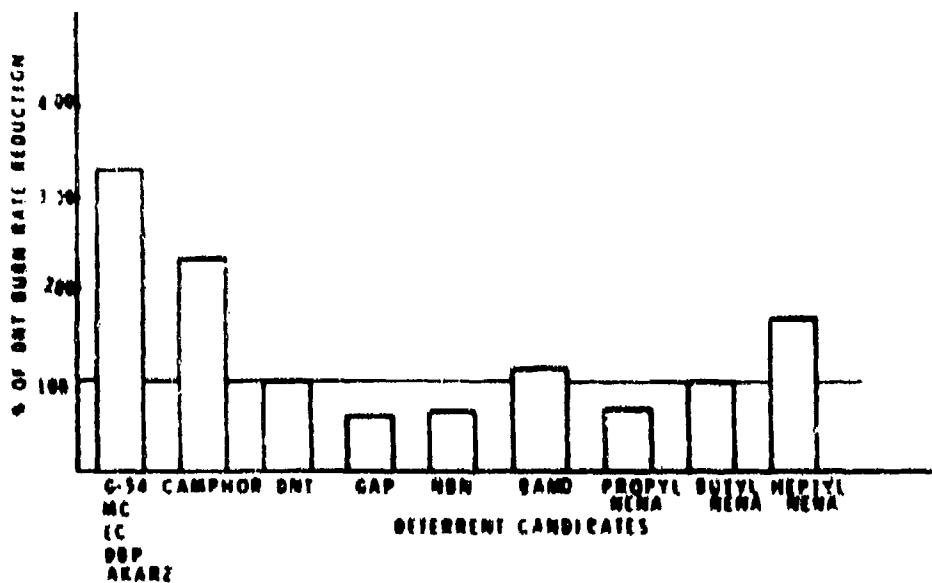


Figure 10. Predicted Burn Rate Reduction of 49 with Various Deterrents (10% Deterrent Concentration)

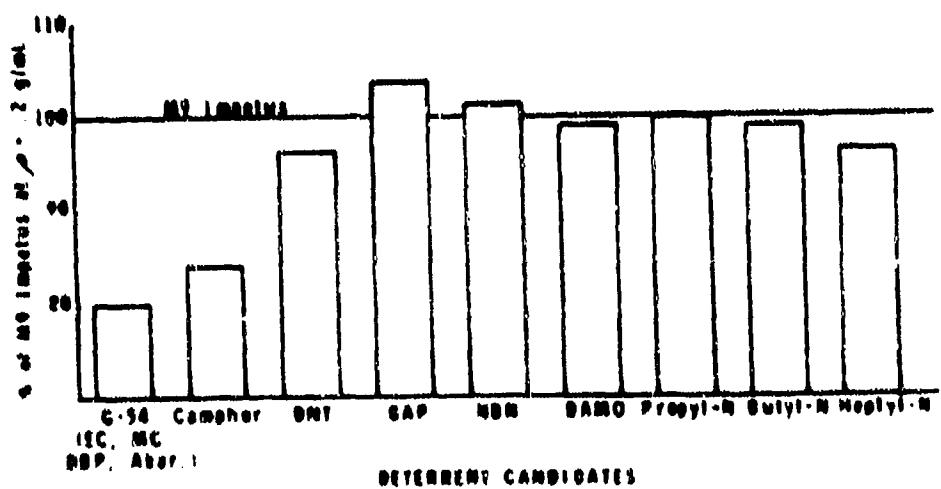


Figure 11. Melting of 49 with various Deterrents at 10% Concentration